REGIO-AND STEREOSELECTIVE PALLADIUM CATALYZED AMINATION OF ALLYLIC SUBSTRATES. SYNTHESIS OF E-4-AMINO-2-ALKEN-1-OL DERIVATIVES

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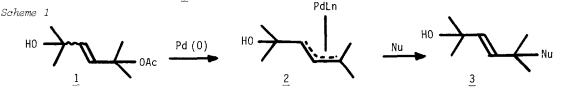
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Summary : Under mild conditions 4, 5 and 6 react with primary and secondary amines in the presence of palladium phosphine complexes as catalysts to give 4-amino-2-alken-1-ols with (E) stereochemistry.

The ability of palladium complexes to catalyze allylic exchange reactions by nucleophiles is an important synthetic methodology heavily documented¹. In addition to carbon nucleophiles the heteronucleophiles and especially those with nitrogen have recently been studied. After the first catalyzed amination described by Atkins² and Takahashi³, these palladium reactions have proved efficient in syntheses of alkaloids⁴, aminosugars⁵, and azaspiranes⁶. These aminations are believed to proceed mainly via π -allyl palladium complexes⁷, and the regio-⁷ and stereochemistry⁸ of the nucleophilic attack by amine depends very much on the reaction conditions.

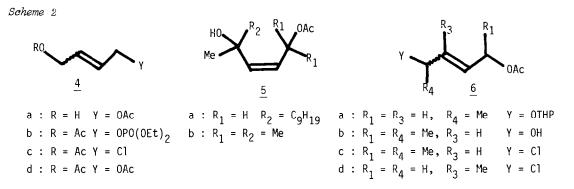
We have previously shown⁹ that π -allyl complexes <u>2</u> generated *chemoselectively* from 1,4-hydroxyacetates <u>1</u> are useful in directing nucleophilic attack by stabilized carbanions. The new carbon-carbon bond was formed exclusively in the 4-position to the hydroxy group, resulting in the formation of compounds <u>3</u> (*Scheme 1*).



The same directing effect has been observed in the palladium (0)-catalyzed alkylation of 1-methoxy-4-acetoxy-2-cyclohexene¹⁰, 1-acetoxy-4-chloro-2-alkenes¹¹, 1,2-epoxy-3-alkenes¹² and dicarboxylates of cyclopentene-1,4-diol¹³. A recent publication¹⁴ also describes the activation of <u>Z</u>-4-acetoxybut-2-enyl dimethylphosphonate <u>4b</u> related to 4-hydroxybut-2-enyl acetate $\underline{4a}^{9c}$ and 4-chlorobut-2-enyl acetate $\underline{4c}^{11}$ (Scheme 2). In the latter publication the question was raised about the *unreactivity* of <u>4a</u> in palladium catalyzed aminations.

We now wish to report our initial observation¹⁵ that <u>Z</u>-4-hydroxybut-2-enyl acetate <u>4a</u> and related 1,4-hydroxyacetates are highly reactive in palladium (0)-catalyzed aminations. We also report that palladium catalyzed amination of chloroacetates <u>6c</u> and <u>6d</u> occurs under mild conditions.



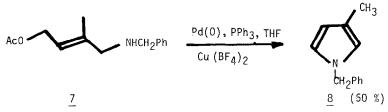


The palladium-catalyzed amination of substrates $\underline{4}$, $\underline{5}$ and $\underline{6}$ works well with both primary and secondary amines (Table 1) and constitutes a general synthesis of 4-amino-2-alken-1-ol derivatives. It is interesting to note that primary amines only give rise to monoalkylated products (entries 4-6 and 9-10)¹⁶. The <u>E</u> and <u>Z</u> diacetate $\underline{4d}$ also underwent a smooth amination reaction and more importantly, the use of one equivalent of Et₂NH in the presence of Et₃N results in amination of *only one* of the two allylic acetoxy groups (entry 2)¹⁷. The new carbon-nitrogen bond was regioselectively formed in the 4-position to the hydroxy or acetoxy group in all cases studied.

The stereochemistry of the 4-amino-2-alken-1-ols was established by 1 H NMR spectroscopy at 200 or 250 MHz. In the cases of disubstituted double bonds the vicinal coupling constants of the olefinic protons J_{HH} = 15 - 16 Hz are consistent only with <u>E</u> configuration of the double bond. In the case of the product obtained from <u>6d</u> (entry 12), mainly one geometrical isomer was observed, which was assigned as the E-isomer¹⁸.

Our studies establish the high reactivity of allylic substrates $\underline{4}, \underline{5}$ and $\underline{6}$ in palladium catalyzed amination. The reactions of these 1,4-hydroxyacetates and the related chloroacetates are highly regio-and stereoselective. The usefulness of the reactions studied is enhanced by the allylic hydroxy or acetoxy group in the products and by the readily available starting materials¹⁹. The products obtained have considerable synthetic utility for cyclization reactions²⁰ and for incorporating a second nucleophile in the allylic oxygen position¹¹The cyclization of the aminoacetate $\underline{7}$ to the pyrrole <u>8</u> demonstrates this point (Scheme 3).

Scheme 3

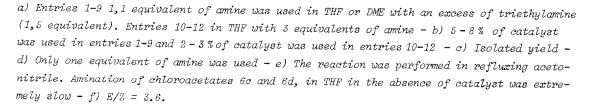


In conclusion, our approach to 4-amino-2-alken-1-ol derivatives appears quite efficient considering the generality and simplicity in synthesis of the starting materials¹⁹.

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entry	substrate	amine	catalyst ^b	cond	itions temp(°C)	products	yield ^C %
1	<u>4a</u> (Z)	Et ₂ NH	Pd(PPh ₃) ₄	0,3	20	На	NEt ₂ 75
2	<u>4d</u> (E) or (Z)	Et_2NH^d	Pd(PPh ₃) ₄	2	20	Ac0	NEt ₂ 70
3	<u>4a</u> (E)		Pd(PPh ₃) ₄	1	20	HO	N 65
4	<u>4a</u> (Z)	PhCH ₂ NH ₂	Pd(PPh ₃) ₄	2	20		NHCH ₂ Ph 65
5	<u>5a</u> (Z)	PhCH ₂ NH ₂	Pd(DIPHOS) ₂	1	₂₅ C		NHCH ₂ Ph 74
6	<u>5b</u> (Z)	PhCH ₂ NH ₂	Pd(DIPHOS) ₂	24	65		< ⁷⁶ NHCH₂Ph
7	<u>6b</u> (Z)	Et ₂ NH	Pd(PPh3)4	4	60	но	NEt _{2 68}
8	<u>6b</u> (Z)	Me PhCH ₂ NH ₂	Pd(PPh ₃) ₄	4	60	но	NHCHMePh 45
9	<u>6a</u> (Z)	PhCH ₂ NH ₂	Pd(DIPHOS) ₂	1	20	но	NHCH2Ph 7521
10	<u>6c</u> (E)	PhCH ₂ NH ₂	Pd(acac) ₂ + 4PPh ₃	9	20	Aco	NHCH ₂ Ph 56
11	<u>6c</u> (E)	Me ₂ NH	Pd(acac) ₂ +4PPh ₃	3	20	Aco	NMe ₂ 80
12	<u>6d</u> (E/Z = 3.6)	Me ₂ NH	Pd(acac) ₂ +4PPh ₃	4	20	Aco	NMe ₂ 71
13	<u>6d</u> (E/Z = 3.6)	PhCH ₂ NH ₂	no catalyst	1,5	80 ^e	Aco	NHCH2Ph 89 ^f

Table 1 : Amination of allylic substrates 4, 5 and 6 with primary or secondary amines^a



References and notes

- a) B.M. Trost; Tetrahedron, 33, 2615 (1977) b) J. Tsuji; Organic Synthesis with Palladium Compounds, Springer Verlag (1980) c) J. Tsuji; Pure and Appl. Chem, 54, 197 (1982) d) B. Bosnich. L.B. Mackenzie; Pure and Appl. Chem. 54, 189 (1982) e) B. Akermark, J.E. Backvall, K. Zetterberg; Act. Chem. Scand. <u>B</u>, <u>36</u>, (9), 577, 1982, and for exhaustive citation see references (9b) and (9c).
- 2. K.E. Atkins, W.E. Walker and R.M. Manyik ; Tetrahedron Letters, 3821 (1970).
- 3. K. Takahashi, A.Miyaki and G. Hata ; Bull. Chem. Soc. Japan, 45, 230 (1972).
- 4. B.M. Trost and J.P. Genêt; J. Amer. Chem. Soc. <u>98</u>, 8516 (1976) B.M. Trost, S.A. Godleski and J.P. Genêt; J. Amer. Chem. Soc, <u>100</u>, <u>3930</u> (1978) - B.M. Trost, S.A. Godleski and J. Belletire; J. Org. Chem., <u>44</u>, 2052 (1979) - R.Z. Andriamialisoa, N. Langlois, Y. Langlois; Heterocycles <u>14</u>, 1457 (1980) - J.E. Bäckvall, R.E. Nordberg, J.E. Nyström, T. Högberg and B. Ulff; J. Org. Chem., <u>46</u>, 3479 (1981).
- H.H. Bear ans Z.S. Hanna ; Carbohydrate Research, <u>78</u> (C₁₁-C₁₄) (1980); Can. J. Chem. <u>59</u>, 889 (1981).
- 6. S.A. Godleski, J.D. Meinart and D.J. Miller; Tetrahedron Letters, 22, 2247 (1981).
- B. Åkermark, G. Åkermark, L.S. Hegedus and K. Zetterberg ; J. Amer. Chem. Soc., <u>103</u>, 3037 (1981).
- 8. a) B. Åkermark, J.E. Bäckvall, A. Löwenborg and K. Zetterberg ; J. Organometal. Chem., <u>166</u> c33 (1979).
 - b) J.E. Bäckvall, R.E. Nordberg, K. Zetterberg and B. Akermark; Submitted for publication.
 - c) B.M. Trost and E. Keinan ; J. Amer. Chem. Soc., <u>100</u>, 7779 (1978).
 - d) B.M. Trost and E. Keinan ; J. Org. Chem., 44, 3451 (1980).
- 9. a) J.P. Genêt, F. Piau, J. Ficini ; Tetrahedron Letters, 21, 3183 (1980).
 - b) J.P. Genêt, F. Piau ; J. Org. Chem. <u>46</u>, 2414 (1981).
 - c) J.P. Genêt, M. Balabane, Y. Legras ; Tetrahedron Letters, <u>23</u>, 331 (1981).
 - d) J.P. Genêt, M. Balabane, F. Charbonnier ; Tetrahedron Letters, 23, 5027 (1982).
- 10. J.E. Bäckvall and R.E. Nordberg ; J. Amer. Chem. Soc., 103, 4959 (1981).
- 11. J.E. Bäckvall, R.E. Nordberg and J.E. Nyström; Tetrahedron Letters, 23, 1617 (1982).
- B.M. Trost and G.A. Molander; J. Amer. Chem. Soc., <u>103</u>, 5969 (1981) J. Tsuji, H. Ueno, Y. Kobayashi and H. Okumoto; Tetrahedron Letters, 22, 2575 (1981).
- 13. R.S. Valpey, D.J. Miller, J.M. Ester and S.A. Godleski ; J. Org. Chem. 47, 4717 (1982).
- 14. Y. Tanigawa, K. Nishimura, A. Kawasaki, S.I. Murahashi; Tetrahedron Letters, 23, 5549 (1982).
- 15. M. Balabane; Thèse de 3ème cycle, june 23 (1982), Université Pierre & Marie Curie (Paris).
- 16. B.M. Trost ; Pure and Appl. Chem. 51, 794 (1979).
- A similar selectivity for monosubstitution was recently observed in the palladium-catalyzed alkylation of cyclopentene-1,4-diol dicarboxylates (réf 13).
- 18. The assignment is based on the fact that the product is identical to the major geometrical isomer obtained from classical nucleophilic substitution of the E/Z mixture (E/Z = 3.6). In the non-catalyzed nucleophilic substitution the olefin geometry should not be affected.
- 19. For the syntheses of hydroxyacetates <u>4a</u>, <u>5b</u>, <u>6a</u> and <u>6b</u> see references (9a-d). The chloroacetates <u>6c</u> and <u>6d</u> were prepared by palladium-catalyzed 1,<u>4</u>-acetoxychlorination of E,Z-2,4-hexadiene <u>and</u> isoprene respectively (11). Hydroxyacetate <u>5a</u> was prepared according to the following sequence :

$$H \rightarrow C \equiv C \rightarrow CH_2OH \xrightarrow{(i)} EtMgBr(2eq.)THF}_{(ii)} Me \xrightarrow{C}_{0} - CgH_{19} \longrightarrow Me \xrightarrow{CgH_{19}}_{OH} C = C \rightarrow CH_2OH \xrightarrow{(i)} H_2, Pd, Lindlar}_{OH} \xrightarrow{5a}_{(ii)} AC_2O, NEt_3, CH_2Cl_2 \xrightarrow{5a}_{(ii)}$$

- J.E. Bäckvall and J.E. Nyström ; J. Chem. Soc. Chem. Comm., 59 (1981) and references cited therein.
- 21. The aminoalcohol was obtained after hydrolysis (HCl, 2 N ; 1 hr, 25°C) of the corresponding tetrahydropyrannyl ether.

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